FTIR/EDX Contaminant Analysis System

for LabSolutions IR

Inquiries regarding contaminant reflect a wide range of issues, including musty odor, chlorinous odor or other abnormal odors and tastes; and the inclusion of contaminants such as various rubber, metal packings and sealing materials. The major cause of contaminants revolves around the aging of the infrastructure and equipment as well as operator's body and clothing surface. While eliminating these causes is difficult, identifying the contaminants as quickly as possible helps to ease the concerns of users.

Infrared spectrometry and X-ray fluorescence spectrometry provide an effective way to identify contaminants.

FTIR/EDX contaminant analysis system is a comprehensive search system that incorporates both techniques.

Procedure for Qualitative Analysis of Contaminants Measure IR spectrum of contaminant. Analyze X-ray fluorescence of contaminant. Search Contaminant Library. Check spectra of similar substances. Select X-ray fluorescence profile. Compare Qualitative analysis of contaminant (Improve precision with combined analysis of organic and inorganic component data.)



Features

- Shimadzu's proprietary contaminant library was prepared with cooperation from organizations in the public water supply industry and food manufactures.
- Includes information from actual contaminant samples.
- Includes both an infrared spectral library and X-ray fluorescence profiles (in PDF file format), which significantly improves precision of contaminant searches.

Applicable Fields

- Food, beverage Petroleum, chemical and material Electrical, electronic
- Machine, transportation Contract research organizations, etc.

Contents

Infrared Spectral Library

Instrument used : Shimadzu Fourier Transform Infrared Spectrophotometer

Measurement method: Single-reflection ATR (prism: Ge or diamond/ZnSe)

Number of spectra : 485 spectra

Information included: Material names, colors, shapes, hardness, etc.

Prism selection criteria: Ge - Enables measurement of samples with high refractive indices

(such as black rubber).

Absorption peak is smaller than diamond/ZnSe.

Diamond/ZnSe - Less prone to scratching and provides larger absorption

peaks than Ge. Cannot measure samples with high refractive

indices (such as black rubber).



Instrument used : Shimadzu Energy Dispersive X-Ray Fluorescence Spectrometer

* The number of available data differs depending on the detection limit of the instruments.

Profile count : 485 profiles

Information included: Qualitative results and FP method quantitative results

(Uses FTIR qualitative results for principal components.)



IRAffinity-1S

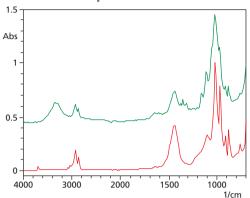


EDX-7000/8000/8100

Example of Infrared Spectral Search Results and Corresponding X-Ray Fluorescence Profile for a Contaminant

The following shows an example of infrared spectral search results and qualitative/quantitative EDX information for matching substances in search results. The contaminant was discovered in a public water system. The Contaminant Library was utilized to determine the contaminant and corresponding profile.

Infrared Spectral Search Results



Detailed information:

Green (measured IR spectrum):

Infrared spectrum from contaminant detected in public drinking water

Red (Contaminant Library):

Interior residue (gray) on fitting seal for 32 mm diameter water drain pipe

: Styrene butadiene styrene (SBS), calcium carbonate (CaCO₃),

and magnesium silicate (talc, Mg3Si4O10(OH)2)

Major element: Cl, Ca, Si, Mg, S, Zn, Ti

Color : Grav Shape : Plastic, ring Hardness : Pliable

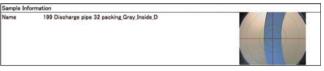
Measurement method: ATR (Diamond)

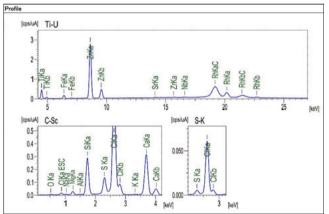
: None

Infrared spectral search results indicated the contaminant is identical or similar to the metal fitting seal material from the water drain pipe (principal component: styrene butadiene styrene [SBS]).

X-Ray Fluorescence Profile of Corresponding Substance

Metal shine





| easurement (| Condition | | | | Collimator | 3mm | Atmosphere | Vac |
|--------------|-----------|-----------|--------|-----------|------------|-----------|------------|-----|
| Analyte | Voltage | Current | Filter | Acr.(keV) | Anal.(keV) | Time | DT% | |
| Ti-U | 50 | 41-Auto | | 0 - 40 | 4.20-37.00 | Live- 100 | 25 | |
| S-K | 15 | 1000-Auto | Al | 0 - 20 | 2.42-2.82 | Live- 300 | 25 | |
| C-Sc | 15 | 318-Auto | | 0 - 20 | 0.10- 4.20 | Live- 300 | 26 | |

| Quantitative Result | | | | | | | | | | |
|---------------------|--------|--------|------------|-----------|------|---------|--|--|--|--|
| Analyte | Result | 120.00 | (Std.Dev.) | ProcCalc. | Line | Int. | | | | |
| CI | 9.114 | % | (0.020) | Quan-FP | CIKa | 0.7206 | | | | |
| Ca | 5.661 | 5 | (0.009) | Quan-FP | CaKa | 4.0402 | | | | |
| Si | 5.391 | % | (0.010) | Quan-FP | SiKa | 2.9346 | | | | |
| Mg | 2.043 | % | (0.015) | Quan-FP | MgKa | 0.2429 | | | | |
| S | 1.154 | % | (0.003) | Quan-FP | S Ka | 1.5087 | | | | |
| Zn | 1.100 | % | (0.003) | Quan-FP | ZnKa | 27.2646 | | | | |
| Ti | 1.054 | % | (0.009) | Quan-FP | TiKa | 3.2267 | | | | |
| Na | 0.956 | % | (0.025) | Quan-FP | NaKa | 0.0398 | | | | |
| Fe | 0.139 | % | (0.002) | Quan-FP | FeKa | 1.4017 | | | | |
| Al | 0.069 | % | (0.003) | Quan-FP | AlKa | 0.0198 | | | | |
| K | 0.039 | % | (0.002) | Quan-FP | K Ka | 0.0202 | | | | |
| Sr | 0.003 | 5 | (0.000) | Quan-FP | SrKa | 0.2143 | | | | |
| Nb | 0.003 | % | (0.000) | Quan-FP | NbKa | 0.2288 | | | | |
| Zr | 0.002 | % | (0.000) | Quan-FP | ZrKa | 0.1738 | | | | |
| C12H14 | 73.272 | % | () | Balance | | | | | | |

Note: Quantitative results were calculated based on the principal components (SBS) determined from infrared spectral search results.

Qualitative and quantitative information can be obtained for iron rust, scaling, and other inorganic components, which is difficult to obtain only with infrared spectra.

It also provides an effective means for qualitative analysis or refining results for seals and other substances with the same principal components (organic substances), but different additives.



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